A FACILE SYNTHESIS OF SUBSTITUTED CONJUGATED CYCLOHEXADIENES.<sup>1</sup>

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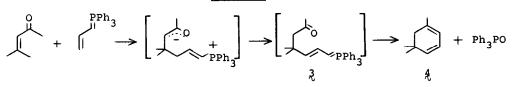
We wish to report a general method for the preparation of cyclohexadienes having structure 1. Previous procedures suffer from lack of specificity and difficulty of execution.<sup>2</sup> Our earlier findings that allylidenetriphenylphos-



 $l, R_1 = alkyl; R_2 - R_5 = alkyl or H 2, n = 1,2,3,4$ 

phoranes add in a conjugate fashion to cyclic enones of type  $2^3$  led us to investigate their reaction with acyclic and semicyclic enones for the preparation of compounds of type  $1.^4$ 

When mesityl oxide was treated with allylidenetriphenylphosphorane in ether, 1,5,5-trimethylcyclohexa-1,3-diene  $(4)^5$  was obtained in 50% yield. In such a reaction, it has been shown that the addition of the  $\gamma$ -position of the ylid to the  $\beta$ -carbon of the enone occurs first, followed by proton transfer to generate ylid 3 which then undergoes an intramolecular Wittig condensation to produce  $4^{3,4}$ (Scheme I). Scheme I

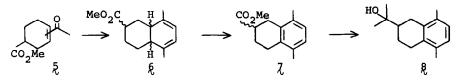


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A variety of  $\alpha$ , $\beta$ -unsaturated ketones were treated in a similar fashion to afford the dienes outlined in Table I.<sup>6</sup>

Application of this diene preparation to the synthesis of the sesquiterpene occidol (a) is outlined in Scheme II. The carbomethoxyacetylcyclohexene 5, prepared by the Friedel-Crafts acylation of methyl cyclohex-3-enecarboxylate,<sup>8</sup> was





converted to  $\xi$  by treatment with 2-butenylidenetriphenylphosphorane in refluxing THF for 4 hr in 30% yield. Subsequent sulphur dehydrogenation of  $\xi$  followed by treatment of the crude  $\gamma$  with ethereal methyllithium afforded racemic occidol<sup>10</sup> in 70% overall yield from  $\xi$ .

If the starting ylid is disubstituted at the  $\gamma$ -position, regeneration of the type 3 is not possible. In one case, when mesityl oxide was condensed with 3-methyl-2-butenylidenetriphenylphosphorane (9), normal 1,2-addition occurred to give a mixture of the corresponding isomeric triene. With 2-methylenecyclohexanone, on the other hand, which is known to be a rather unusually reactive Michael acceptor, conjugate addition of the Wittig reagent 9 still occurs; now, however, from the  $\alpha$ -position of the ylid to give the betaine 10, which subsequently closes to the cyclopropane 11, (Scheme III). Infrared absorption at 1700 cm<sup>-1</sup>, nmr (CC1<sub>4</sub>)  $\delta$  4.73 (br d, 1), 1.55-2.40 (m, 16), 0.40 (m, 1), uv  $\lambda_{max}$  237 nm ( $\epsilon$  = 5500), and parent ion 178 support the assigned structure 11. A few cases of cyclopropane formation have been reported for the addition of simple methyl-enetriphenylphosphoranes with unsaturated carbonyl compounds.<sup>11</sup> This result is, however, in contrast to a similar condensation of 2-methylenecyclohexanone with (2-cyclohexylidene-ethylidene)triphenylphosphorane, which was reported to give

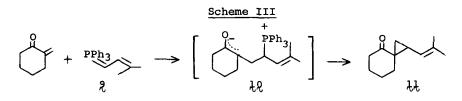


Table I. Products and Yields of Ylid Reactions.				
Enone	Ylida	Diene Product	Conditions <sup>C</sup>	Yield, % <sup>d</sup>
<i>∕</i> ₩	A PPh3	$\diamond$	Ether, 25°, 30 hr	62
$\succ$	≫ <sub>pph3</sub>	$\Rightarrow$	Ether, 25°, 30 hr	50 <sup>e</sup>
° 7	PPh3		THF, 25°, 18 hr	25 <sup>f</sup>
	≫ <sub>PPh3</sub>	Ŭ	THF, 25°, 16 hr	41 <sup>g</sup>
OL	Pph3		THF, reflux, 30 hr	(25) <sup>h</sup>
Meo <sub>2</sub> c 9	PPh3	MeO <sub>2</sub> C	THF, reflux, 4 hr	(30) <sup>i</sup>
	≫_ PPh <sub>3</sub>	$\searrow$	Ether, 25°, 12 hr	32

1-(1-cyclohexen-1-y1)-3-cyclohexylidenepropene via a [1,3]-hydrogen shift.<sup>12</sup>

<sup>a)</sup>Ylids generated from phosphonium bromides with <u>n</u>-butyllithium. <sup>b)</sup>Spectra and elemental analyses were consistant with the assigned structures. The dienes were the only volatile products obtained unless indicated otherwise. <sup>c)</sup>At least a 15% excess of ylid was used. For the volatile dienes ether was chosen to faci litate product isolation. <sup>d)</sup>Figures in parentheses refer to vpc yields; other figures refer to isolated yields. Yields were not optimized. <sup>e)</sup>Yield was 83% based on recovered starting material. <sup>f)</sup>The yield was low as a result of dimerization of the 2-methylenecyclohexanone. <sup>g)</sup>Pulgenone dimer accounted for the remainder of the starting material. <sup>h)</sup>Two isomeric trienes were also obtained in yields of 26 and 16%. <sup>i)</sup>A 60% yield of volatile product was obtained. Vpc analysis indicates five products and nmr analysis showed that some trienes were present.

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